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Key indicators

Single-crystal X-ray study
T = 296 K
Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$
R factor = 0.044
wR factor = 0.108
Data-to-parameter ratio = 12.6

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

A new key intermediate in the synthesis of 6-oxopiperidine-2-carboxylate derivatives

The relative stereochemistry of methyl 6-oxo-5-(5-phenyl-tetrazol-2-yl)piperidine-2-carboxylate, $\text{C}_{14}\text{H}_{15}\text{N}_5\text{O}_3$, has been determined. It confirms the *cis* configuration of the piperidine ring as well as the position of the substituent on the tetrazole ring. The packing of the molecules is influenced by $\text{N}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds.

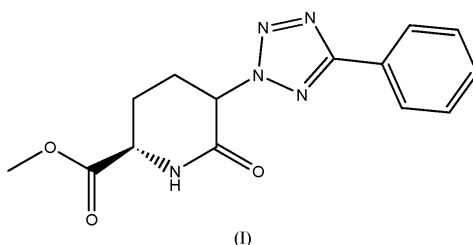
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Comment

Pipecolic acid derivatives are an important class of compounds, which can be used as starting materials for several synthetic drugs, such as enzyme inhibitors (Perumattam *et al.*, 1991), immunosuppressors (Jones *et al.*, 1989), antibiotics (Sehgal *et al.*, 1983) and mycotoxic agents (Martens & Scheunemann, 1991). With the aim of developing new tetrazolic alpha-iminoacids derived from pipecolic acid, an analogue of *cis*-4-(tetrazolylalkyl)piperidine-2-carboxylic acid, giving selective and potential antagonistic activity at the *N*-methyl-D-aspartic acid receptor (NMDA; Orstein *et al.*, 1991), we have prepared a key intermediate, (I), starting from *meso*-dimethyldibromoadipate (Guha & Sankaran, 1955).



The double substitution of the dibromo derivative successively by 5-phenyl-1*H*-tetrazole and the azide group followed by catalytic hydrogenation led, *via* an intramolecular aminolysis, to the formation of a racemic mixture of two diastereoisomers (*SS*, *RR* and *SR*, *RS*) with a diastereoisomeric ratio of 5/1. The presence and the ratio of the two diastereoisomers were determined by ¹H NMR (300 MHz) analysis of the mixture, but determination of the relative stereochemistry of the piperidine core by this method was not possible. The major diastereoisomer was crystallized in a diethyl ether/ethyl acetate mixture, analysed by X-ray diffraction and shown to contain the *SS* and *RR* isomers. Consequently, the relative position of the substituents on the heterocyclic core is *cis* (Fig. 1). The formation of this isomer can be explained by successive $\text{S}_{\text{N}}2$ reactions of the Br atoms of the *meso*-dimethyldibromoadipate, which, after cyclization, would orient the two substituents in a *cis* position (diastereoisomer *SS*, *RR*). Nevertheless, this control is not total since some *trans* isomer

(diastereoisomer *SR*, *RS*) was obtained. Furthermore, the X-ray analysis shows that the first substitution, which corresponds to the alkylation by the 5-phenyl-1*H*-1-tetrazole, occurred only at the N2 position of the tetrazole ring. The angles between the C1–C13 and N2–C3 bonds are 148.6 (2) and 145.9 (2)°. The angle between the phenyl ring and the tetrazole ring is 13.2 (1)°. The crystal cohesion is assured by van der Waals interactions and hydrogen bonding (Fig. 2). Thus, atom N1 forms a hydrogen bond with atom O3(1 – *x*, –*y*, –*z*), and atom C3 forms a bond with atom N4(*x*, $\frac{1}{2}$ – *y*, $-\frac{1}{2}$ + *z*) (Table 1).

Experimental

The title compound was synthesized by successive alkylation of 5-phenyltetrazole in the presence of triethylamine and the action of sodium azide on *meso*-dimethyldibromoadipate, followed by catalytic hydrogenation over Pd/C. The crystallization was carried out at ambient temperature with a mixture of diethyl ether and ethyl acetate.

Crystal data

C ₁₄ H ₁₅ N ₅ O ₃	<i>Z</i> = 4
<i>M_r</i> = 301.31	<i>D_x</i> = 1.356 Mg m ^{–3}
Monoclinic, <i>P</i> 2 ₁ / <i>c</i>	Mo <i>K</i> α radiation
<i>a</i> = 12.7980 (6) Å	<i>θ</i> = 1–25
<i>b</i> = 10.8334 (3) Å	<i>μ</i> = 0.10 mm ^{–1}
<i>c</i> = 10.9142 (3) Å	<i>T</i> = 296 (2) K
<i>β</i> = 102.796 (1)°	Block, colourless
<i>V</i> = 1475.63 (9) Å ³	0.10 × 0.08 × 0.05 mm

Data collection

Nonius KappaCCD diffractometer	<i>R</i> _{int} = 0.020
<i>ω</i> scans	<i>θ</i> _{max} = 25.0°
Absorption correction: none	<i>h</i> = –15 → 14
5170 measured reflections	<i>k</i> = –12 → 12
2563 independent reflections	<i>l</i> = –12 → 12
1777 reflections with <i>I</i> > 2σ(<i>I</i>)	

Refinement

Refinement on <i>F</i> ²	$w = 1/[\sigma^2(F_o^2) + (0.0438P)^2 + 0.1856P]$
$R[F^2 > 2\sigma(F^2)] = 0.044$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.108$	(Δ/σ) _{max} = 0.001
<i>S</i> = 1.04	$\Delta\rho_{max} = 0.16 \text{ e } \text{Å}^{-3}$
2563 reflections	$\Delta\rho_{min} = -0.15 \text{ e } \text{Å}^{-3}$
204 parameters	Extinction correction: <i>SHELXL97</i>
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: 0.026 (4)

Table 1

Hydrogen-bonding geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
N1–H1N···O3 ⁱ	0.90 (2)	1.99 (2)	2.880 (2)	175
C3–H3···N4 ⁱⁱ	0.98	2.57	3.528 (2)	167

Symmetry codes: (i) 1 – *x*, –*y*, –*z*; (ii) $\frac{1}{2}$ – *y*, *z* – $\frac{1}{2}$.

All H atoms were located in difference maps, but H atoms attached to C atoms were thereafter treated as riding atoms, with C–H distances of 0.98 (CH₃), 0.97 (CH₂) and 0.93 (C–H aromatic) Å and *U*_{iso}(H) = 1.2*U*_{eq}(C) or 1.5*U*_{eq}(C_{methyl}) [N–H1 = 0.89 (2) Å]. The N–H bond length was restrained to 0.89 (2) Å.

Data collection: *KappaCCD Server Software* (Nonius, 1998); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997); data

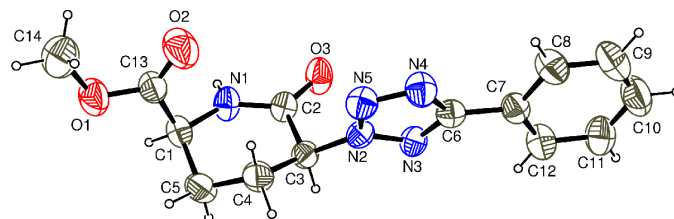


Figure 1

A view of (I), with the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level.

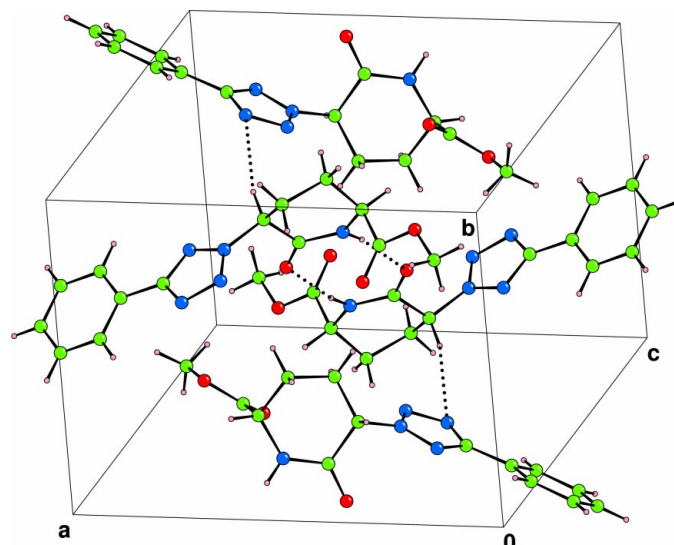


Figure 2

A view of the crystal packing, showing N–H···O and C–H···N intermolecular hydrogen interactions as dashed lines.

reduction: *DENZO* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *CAMERON* (Watkin *et al.*, 1993); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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